

First Example of Alkynyliodonium Tosylates coupling with 1,1-Bimetalloalkenes of Selenium and Zirconium†

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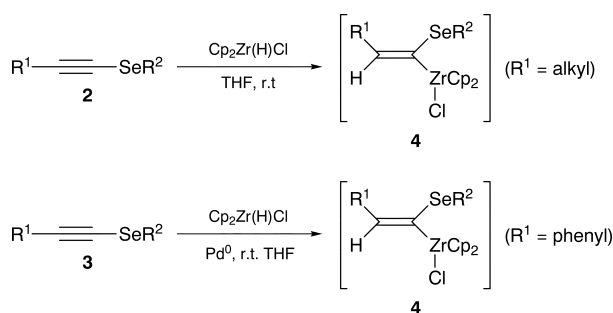
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Hydrozirconation of internal acetylenic selenides afforded 1,1-bimetalloalkenes, (*E*)- α -selanylvinylzirconium, which can cross-couple with alkynyliodonium tosylates directly in the presence of Pd(PPh₃)₄ as the catalyst.

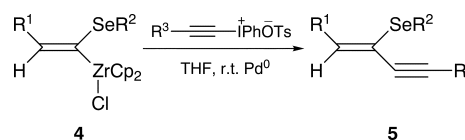
The stereoselective synthesis of conjugated enynes is of great importance in organic chemistry because of many natural compounds containing their structural skeletons.¹ A number of methods for the preparation of conjugated enynes have been previously reported, particularly transition-metal-catalyzed processes.¹ Coupling of stereodefined alkenyl metallic compounds include vinylboron,² vinylcopper,^{3,4} vinylzinc,^{5,6} vinylaluminum⁷ or vinylmagnesium reagents⁸ with haloalkynes. Recently, the synthesis of conjugated functionalized enynes have received more attention. Although various methods for the synthesis of vinylic selenides have been studied intensively,⁹ the synthesis of 1,3-enynylselenides has been little explored.¹⁰ In a previous paper, we reported the coupling of diaryliodonium salts and vinylzirconium compounds in the presence of Pd(PPh₃)₄,¹¹ compared with the cross-coupling of vinylzirconium compounds with arylhalides, which has the advantages of lower reaction time and mild reaction conditions. As an extension of our studies, herein we report the stereoselective formation of selanyl-substituted 1,3-enynes *via* a new carbon-carbon bond-forming reaction, involving the coupling of alkynyliodonium tosylates **1** and (*E*)- α -selanylvinyl zirconium compounds **4** in the presence of Pd(PPh₃)₄.

Alkynyliodonium tosylates **1** are readily available in reasonable yields by interaction of Koser's reagent, PhI(OH)OTs, with terminal alkynes.¹² (*E*)- α -Selanylvinyl zirconium compounds **4** are readily made by hydrozirconation of internal acetylenic selenides **2**, **3**; hydrozirconation of phenylacetylenic selenides **3** can not be directly performed even over extended times or at a raised temperature, however, in the presence of 5% mmol Pd(PPh₃)₄, **3** can react with Cp₂Zr(H)Cl rapidly in about 1 h at room temperature (Scheme 1). Alkynyliodonium tosylate **1** was added to a mixture of (*E*)- α -selanylvinyl zirconium compound **4** and 5% mmol Pd(PPh₃)₄ in THF and stirred for about 3 h, which resulted in the stereoselective synthesis of conjugated enynylselenides **5** (Scheme 2).

As the data in Table 1 show, this method affords a variety of 1,3-enynylselenides stereoselectively in good

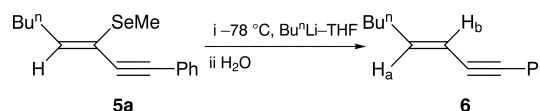


Scheme 1



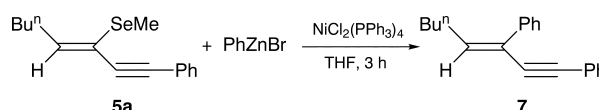
Scheme 2

yields, and more importantly, the reaction can be performed at room temperature in about 2 h. The configuration of selenoenyne **5a** could be confirmed from compound **6** which was obtained by treatment of **5a** with *n*-butyllithium in THF followed by hydrolysis; the reaction occurs stereoselectively (Scheme 3).¹³ Particularly diagnostic for the stereochemistry of **6** was the coupling constant between the vicinal protons H_a and H_b which show a typical value of J_{HH} of 16 Hz which is consistent with an *E* configuration.



Scheme 3

Conjugated enynylselenides are important synthetic intermediates and are effective precursors for synthesizing conjugated enynes, for example, **5a** at room temperature in THF can be coupled with PhZnBr in the presence of 5 mmol% NiCl₂(PPh₃)₂ for 3 h to give **7** in an isolated yield of 60% (Scheme 4).



Scheme 4

In summary, we have combined vinylzirconium compounds and hypervalent iodine salts to synthesize conjugated enynes. Compared to other methods that have been reported, this method has the advantages of mild conditions, simple procedure and lower reaction time.

Experimental

¹H NMR spectra were recorded on an AZ-300 MHz with TMS as internal standard. Mass spectra were determined using a Finigan 8230 mass spectrometer. IR spectra were obtained in neat capillary cells on a Shimadzu IR-408 instrument. The reactions were carried out in pre-dried glassware (150 °C, 4 h) and cooled under a stream of dry nitrogen. All solvents were dried, deoxygenated and redistilled before use.

General Procedure for the Synthesis of 1,3-Enynylselenides 5a-f.—To a freshly prepared suspension of Cp₂Zr(H)Cl (2 mmol) in THF (15 ml) at r.t. was injected acetylenic selenides **2** (2 mmol) in THF (2 ml) [for added phenylacetylenic selenides **3**, Pd(PPh₃)₄ (2 × 5% mmol) must be added simultaneously]. The reaction mixture was stirred for *ca.* 30 min and turned green, then Pd(PPh₃)₄ (2 × 5% mmol) and alkynyliodonium tosylates **1** (2 mmol) in THF (2 ml) were added to the solution at r.t. and stirred for 2 h. The

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Table 1 Alkynylidodonium/(*E*)- α -selanylvinylzirconium compound couplings

Entry	Alkynylidodonium salt	(<i>E</i>)- α -Selanylvinylzirconium	Enzyne	Yield ^a (%)
5a	Ph—C≡C— ⁺ I ⁻ PhO ⁻ Ts			80
5b	Ph—C≡C— ⁺ I ⁻ PhO ⁻ Ts			75
5c	Ph—C≡C— ⁺ I ⁻ PhO ⁻ Ts			85
5d	Ph—C≡C— ⁺ I ⁻ PhO ⁻ Ts			83
5e	MeOCH2—C≡C— ⁺ I ⁻ PhO ⁻ Ts			75
5f	MeOCH2—C≡C— ⁺ I ⁻ PhO ⁻ Ts			70

^aIsolated yield. ^bIn the presence of Pd(PPh₃)₄.

product was washed with saturated aq. NH₄Cl (10 ml) then extracted into diethyl ether, dried with MgSO₄, filtered and concentrated *in vacuo*. The residue was purified by flash chromatography on a 3 ft × 1 in column with light petroleum (bp 30–60 °C) eluent to give **5a–f**.

5a: ¹H NMR δ 7.70–7.20 (m, 5H), 6.10 (t, *J* 6.6 Hz, 1H), 2.18 (s, 3H), 1.96–2.34 (m, 2H), 1.14–1.49 (m, 4H), 0.81 (t, 3H). MS: *m/z* 278 (M⁺+1) (Found: C, 64.93; H, 6.44. C₁₅H₁₈Se requires C, 64.98; H, 6.54%).

5b: ¹H NMR δ 7.69–7.18 (m, 5H), 6.08 (t, *J* 6.6 Hz, 1H), 2.53 (q, 2H), 1.94–2.30 (m, 2H), 1.47 (t, 3H), 1.12–1.50 (m, 4H), 0.79 (t, 3H). MS: *m/z* 292 (M⁺+1) (Found: C, 65.65; H, 6.46. C₁₆H₂₀Se requires C, 65.97; H, 6.92%).

5c: ¹H NMR δ 7.65–7.10 (m, 5H), 6.10 (t, *J* 6.7 Hz, 2H), 3.80 (d, 2H), 3.24 (s, 3H), 2.17 (s, 3H), MS: *m/z* 265 (M⁺+1) (Found: C, 58.53; H, 5.02. C₁₃H₁₄SeO requires C, 58.87; H, 5.32%).

5d: ¹H NMR δ 7.66–7.12 (m, 5H), 6.12 (t, *J* 6.7 Hz, 1H), 3.83 (d, 2H), 3.25 (s, 3H), 2.55 (q, 2H), 1.52 (t, 3H). MS: *m/z* 279 (M⁺+1) (Found: C, 59.74; H, 5.35. C₁₄H₁₆SeO requires C, 60.22; H, 5.78%).

5e: ¹H NMR δ 7.50–7.0 (m, 5H), 6.78 (s, 1H), 4.12 (s, 2H), 3.29 (s, 3H), 2.15 (s, 3H), MS: *m/z* 266 (M⁺+1) (Found: C, 58.40; H, 4.86. C₁₃H₁₄SeO requires C, 58.88; H, 5.32%).

5f: ¹H NMR δ 7.49–7.02 (m, 5H), 6.77 (s, 1H), 4.14 (s, 2H), 3.30 (s, 3H), 2.65 (q, 2H), 1.49 (t, 3H). MS: *m/z* 280 (M⁺+1) (Found: C, 59.82; H, 5.40. C₁₄H₁₆SeO requires C, 60.22; H, 5.78%).

Addition of BuⁿLi to 1,3-Enynylselenide 5a → Enzyne 6.—BuLi (1.6 M hexane solution, 1.1 mmol) was added to a THF (5.0 ml) solution of **5a** (1.0 mmol) at –78 °C. After stirring for 30 min, the mixture was hydrolyzed with saturated aq. NH₄Cl and extracted with CH₂Cl₂ (2 × 10 ml). The organic extract was dried with MgSO₄, filtered and concentrated *in vacuo*. The residue was purified by column chromatography over silica gel (hexane as eluent) to give **6** (yield: 90%). ¹H NMR δ 7.24 (m, 5H), 6.17 (dt, 1H, *J* 6.5 and 16 Hz), 5.62 (d, 1H, *J* 16 Hz), 2.15 (q, 2H), 1.20–2.05 (m, 4H), 0.93 (t, 3H).

General Procedure for the Synthesis of 7.—PhZnBr (1.0 mmol) in THF (3.0 ml) was added to a THF (5.0 ml) solution of selenoenyne **5a** (1.0 mmol) and NiCl₂(PPh₃)₂ (1.0 × 5% mmol) at r.t. After stirring for 3 h, the mixture was washed with saturated aq. NH₄Cl. The product was extracted with hexane and dried over MgSO₄, filtered and concentrated *in vacuo*. The residue was purified by flash chromatography on a 3 ft × 1 in column with hexane as eluent to give **8**. ¹H NMR δ 7.71–7.01 (m, 10H), 6.30 (t, 1H, *J* 6.8 Hz), 2.45–2.10 (m, 2H), 1.43–1.22 (m, 4H), 0.84 (t, 3H).

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References

- T. Tokuyama, K. Uenoyama, G. Brown, J. W. Daly and B. Witkop, *Helv. Chim. Acta.*, 1974, **57**, 2597 and references therein; R. Fujimoto, Y. Kishi and J. F. Blount, *J. Am. Chem. Soc.*, 1980, **102**, 7154; G. R. Schulte, M. C. G. Chung and P. J. Scheuer, *J. Org. Chem.*, 1981, **46**, 3870; A. Guerrero, F. Camps, J. Coll, M. Riba, J. Einhorn, Ch. Descoins and J. Y. Lallemond, *Tetrahedron Lett.*, 1981, 2013.
- N. Miyaoura, K. Yamada, H. Sugimoto and A. Suzuki, *J. Am. Chem. Soc.*, 1985, **107**, 972.
- J. F. Normant, A. Commercon and J. Villieras, *Tetrahedron Lett.*, 1975, 1465; A. Pelter, K. Smith and M. Tabata, *J. Chem. Soc., Chem. Commun.*, 1975, 857.
- I. Rhee, M. Ryang and S. Tsutsumi, *Tetrahedron Lett.*, 1969, 4593; A. Alexakis, G. Cahiez and J. F. Normant, *Synthesis*, 1979, 826.
- A. O. King, N. Okukada and E. I. Negishi, *J. Chem. Soc., Chem. Commun.*, 1977, 683.
- P. A. Magriotis, M. E. Scott and K. D. Kim, *Tetrahedron Lett.*, 1991, **32**, 6085; M. Abarbri, J. L. Parrain, J. C. Cintrat and A. Duchene, *Synthesis*, 1996, 82.
- S. Negishi and A. O. King, *J. Chem. Soc., Chem. Commun.*, 1976, 17; M. Kobayashi, L. F. Valente and E. Negishi, *Synthesis*, 1980, 1034.
- H. P. Dang and G. Linstrumelle, *Tetrahedron Lett.*, 1978, 191.
- (a) D. Y. Yang and X. Huang, *J. Organomet. Chem.*, 1996, 523, 139; (b) D. Y. Yang and X. Huang, *Tetrahedron Lett.*, 1997, 62; (c) D. Y. Yang and X. Huang, *J. Chem. Res. (S)*, 1996, 470; (d) D. Y. Yang and X. Huang, *Synth. Commun.*, 1996, **26**, 4369.
- (a) J. V. Coamsseto and C. A. Brandt, *Synthesis*, 1987, 146; (b) L. S. Zhu, Z. Z. Huang and X. Huang, *Tetrahedron*, 1996, **52**, 9819.
- X. Huang and A. M. Sun, *Synth. Commun.*, 1998, **28**, 773.
- P. J. Stang, B. W. Surber, Z. C. Chen and K. A. Roberts, *J. Am. Chem. Soc.*, 1987, **109**, 228.
- B. T. Grobel and D. Seebach, *Chem. Ber.*, 1977, **110**, 867.